

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

***N'*-Cyclododecylidenepyridine-4-carbohydrazide**Andreas Lemmerer,^{a*} Joel Bernstein^b and Volker Kahlenberg^c^aMolecular Sciences Institute, School of Chemistry, University of the Witwatersrand, Johannesburg, PO Wits 2050, South Africa, ^bFaculty of Science, NYU Abu Dhabi, PO Box 129188, Abu Dhabi, United Arab Emirates, and ^cInstitute of Mineralogy and Petrography, University of Innsbruck, Innsbruck 6020, Austria

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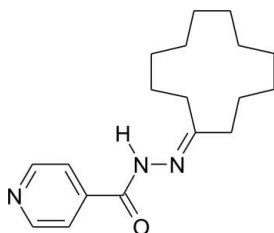
Received 7 March 2012; accepted 19 March 2012

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.111; data-to-parameter ratio = 14.4.

The title compound, $\text{C}_{18}\text{H}_{27}\text{N}_3\text{O}$, is a derivative of the antituberculosis drug isoniazid (systematic name: pyridine-4-carbohydrazide). The crystal structure consists of repeating $C(4)$ chains along the b axis, formed by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds with adjacent amide functional groups that are related by a b -glide plane. The cyclododecyl ring has the same approximately 'square' conformation, as seen in the parent hydrocarbon cyclododecane.

Related literature

For hydrogen-bonding motifs, see: Bernstein *et al.* (1995). For cycloalkane ring conformations, see: Dale (1966).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{27}\text{N}_3\text{O}$
 $M_r = 301.43$
 Orthorhombic, $Pbca$
 $a = 14.8450$ (6) Å

$b = 8.0980$ (4) Å
 $c = 27.3910$ (11) Å
 $V = 3292.8$ (2) Å³
 $Z = 8$

Cu $K\alpha$ radiation
 $\mu = 0.60$ mm⁻¹

$T = 100$ K
 $0.44 \times 0.34 \times 0.2$ mm

Data collection

Oxford Diffraction Xcalibur Ruby Gemini ultra diffractometer
 Absorption correction: multi-scan (*CrysAlis PRO*; Oxford Diffraction, 2006)
 $T_{\min} = 0.779$, $T_{\max} = 0.890$

23848 measured reflections
 2931 independent reflections
 2535 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.048$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.111$
 $S = 1.08$
 2931 reflections
 203 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.17$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.88 (2)	2.15 (2)	3.0122 (15)	164.7 (16)

Symmetry code: (i) $-x + \frac{3}{2}, y + \frac{1}{2}, z$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2006); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

This work was supported in part by grant No. 2004118 from the United States–Israel Binational Science Foundation (Jerusalem). AL thanks the South African National Research Foundation for a postdoctoral scholarship (SFP2007070400002) and the Oppenheimer Memorial Trust for financial support, and the Molecular Sciences Institute for infrastructure support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FJ2532).

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supplementary materials

Acta Cryst. (2012). E68, o1205 [doi:10.1107/S1600536812011774]

N'*-Cyclododecylidenepyridine-4-carbohydrazide*Andreas Lemmerer, Joel Bernstein and Volker Kahlenberg****Comment**

Fig. 1 shows the atomic numbering scheme of the title compound. The amide functional groups form a torsion angle of 38.54 (17)° with the pyridine ring. Fig. 2 shows the *C*(4) (Bernstein *et al.*, 1995) hydrogen bonded ring formed with adjacent amide functional groups, leading to a chain along the *b*-axis. The cyclododecyl ring has a square conformation, as seen in the related cycloalkane C₁₂H₂₄ ring (Dale, 1966).

Experimental

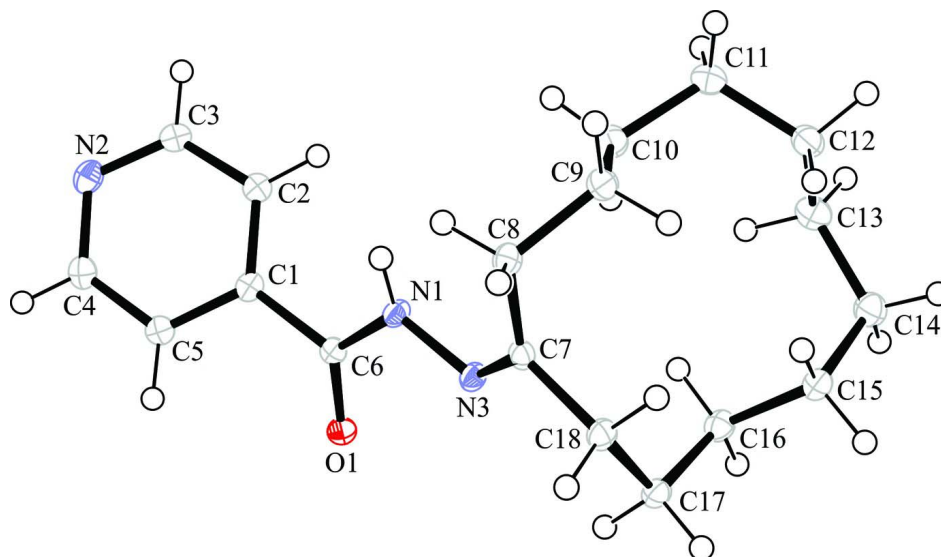
A stoichiometric amount in the ratio of 1:1 of isonicotinic acid hydrazide to cyclododecanone was dissolved in 5 ml of methanol. The solution was refluxed for a few hours, and left to cool to room temperature. Colourless, block-like crystals were harvested after slow evaporation over a few days at ambient conditions.

Refinement

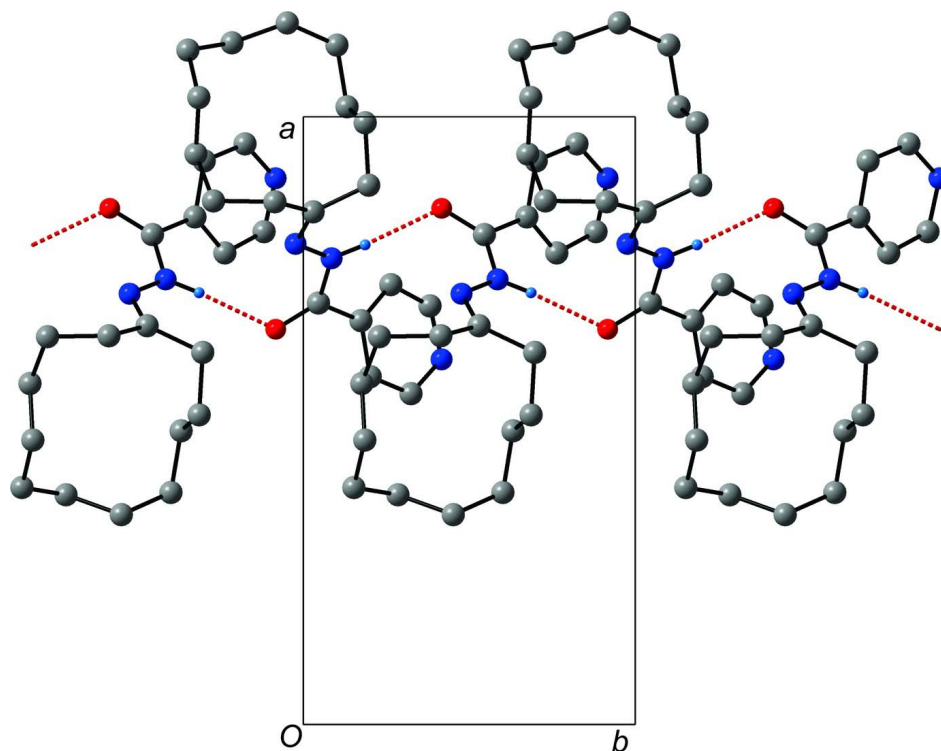
The C-bound H atoms were geometrically placed (C—H bond lengths of 0.95 (aromatic CH) and 0.99 (methylene CH₂) Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The N-bound H atoms were located in the difference map and coordinates refined freely together with their isotropic thermal parameters.

Computing details

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2006); cell refinement: *CrysAlis PRO* (Oxford Diffraction, 2006); data reduction: *CrysAlis PRO* (Oxford Diffraction, 2006); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

**Figure 1**

The asymmetric unit of (I) showing the atomic numbering scheme. Displacement ellipsoids are shown at the 50% probability level.

**Figure 2**

Hydrogen bonding chain showing the C(4) hydrogen bonded chains. Intermolecular N—H...O hydrogen bonds are shown as dashed red lines.

***N'*-Cyclododecylidenepyridine-4-carbohydrazide**

Crystal data

$C_{18}H_{27}N_3O$	$F(000) = 1312$
$M_r = 301.43$	$D_x = 1.216 \text{ Mg m}^{-3}$
Orthorhombic, <i>Pbca</i>	Cu <i>K</i> α radiation, $\lambda = 1.5418 \text{ \AA}$
Hall symbol: -P 2ac 2ab	Cell parameters from 9464 reflections
$a = 14.8450 (6) \text{ \AA}$	$\theta = 3.0\text{--}67.5^\circ$
$b = 8.0980 (4) \text{ \AA}$	$\mu = 0.60 \text{ mm}^{-1}$
$c = 27.3910 (11) \text{ \AA}$	$T = 100 \text{ K}$
$V = 3292.8 (2) \text{ \AA}^3$	Block, colourless
$Z = 8$	$0.44 \times 0.34 \times 0.2 \text{ mm}$

Data collection

Oxford Diffraction Xcalibur Ruby Gemini ultra diffractometer	2931 independent reflections
ω scans	2535 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (<i>CrysAlis PRO</i> ; Oxford Diffraction, 2006)	$R_{\text{int}} = 0.048$
$T_{\text{min}} = 0.779$, $T_{\text{max}} = 0.890$	$\theta_{\text{max}} = 67.0^\circ$, $\theta_{\text{min}} = 3.2^\circ$
23848 measured reflections	$h = -17 \rightarrow 16$
	$k = -9 \rightarrow 9$
	$l = -32 \rightarrow 32$

Refinement

Refinement on F^2	H atoms treated by a mixture of independent and constrained refinement
Least-squares matrix: full	$w = 1/[\sigma^2(F_o^2) + (0.0722P)^2 + 0.6785P]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.111$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
2931 reflections	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
203 parameters	
0 restraints	

Special details

Experimental. Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm in *CrysAlisPro*.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.83538 (9)	0.67344 (15)	0.71252 (5)	0.0140 (3)
C2	0.77610 (9)	0.76355 (16)	0.74142 (5)	0.0151 (3)
H2	0.713	0.7447	0.7397	0.018*
C3	0.81105 (9)	0.88164 (16)	0.77286 (5)	0.0171 (3)
H3	0.7701	0.9417	0.7927	0.02*
C4	0.95549 (9)	0.82425 (17)	0.74991 (5)	0.0185 (3)
H4	1.0183	0.8447	0.7527	0.022*
C5	0.92753 (9)	0.70136 (16)	0.71813 (5)	0.0157 (3)
H5	0.9702	0.6374	0.7005	0.019*
C6	0.80631 (9)	0.54546 (16)	0.67606 (4)	0.0131 (3)

C7	0.65632 (8)	0.52936 (16)	0.57794 (5)	0.0144 (3)
C8	0.61244 (9)	0.69684 (16)	0.57370 (5)	0.0160 (3)
H8A	0.6251	0.7436	0.541	0.019*
H8B	0.6386	0.7721	0.5984	0.019*
C9	0.50972 (9)	0.68614 (17)	0.58126 (5)	0.0179 (3)
H9A	0.4828	0.7957	0.5746	0.022*
H9B	0.4843	0.607	0.5574	0.022*
C10	0.48338 (9)	0.63169 (18)	0.63275 (5)	0.0208 (3)
H10A	0.5026	0.7177	0.6562	0.025*
H10B	0.5163	0.5289	0.6408	0.025*
C11	0.38197 (9)	0.60107 (18)	0.63905 (5)	0.0221 (3)
H11A	0.3691	0.5856	0.6742	0.027*
H11B	0.349	0.7008	0.6281	0.027*
C12	0.34546 (9)	0.45158 (17)	0.61105 (5)	0.0202 (3)
H12A	0.2788	0.4539	0.6123	0.024*
H12B	0.3636	0.4613	0.5764	0.024*
C13	0.37813 (10)	0.28492 (17)	0.63069 (5)	0.0219 (3)
H13A	0.4364	0.3016	0.6475	0.026*
H13B	0.3343	0.2447	0.6552	0.026*
C14	0.39013 (10)	0.15214 (17)	0.59157 (6)	0.0224 (3)
H14A	0.3336	0.1432	0.5726	0.027*
H14B	0.4004	0.0446	0.6078	0.027*
C15	0.46814 (9)	0.18511 (17)	0.55621 (5)	0.0190 (3)
H15A	0.4653	0.1034	0.5294	0.023*
H15B	0.4601	0.2961	0.5417	0.023*
C16	0.56119 (9)	0.17631 (16)	0.57975 (5)	0.0178 (3)
H16A	0.5722	0.0615	0.5906	0.021*
H16B	0.5615	0.2476	0.6091	0.021*
C17	0.63832 (9)	0.22928 (17)	0.54617 (5)	0.0178 (3)
H17A	0.6962	0.1979	0.5615	0.021*
H17B	0.6333	0.1683	0.515	0.021*
C18	0.63976 (9)	0.41420 (17)	0.53531 (5)	0.0168 (3)
H18A	0.687	0.4349	0.5105	0.02*
H18B	0.5813	0.4443	0.5204	0.02*
N1	0.73263 (7)	0.58502 (14)	0.64919 (4)	0.0148 (3)
H1	0.7106 (12)	0.686 (2)	0.6498 (6)	0.024 (4)*
N2	0.89901 (8)	0.91633 (14)	0.77697 (4)	0.0191 (3)
N3	0.70898 (7)	0.47642 (14)	0.61174 (4)	0.0148 (3)
O1	0.84951 (6)	0.41677 (11)	0.67137 (3)	0.0159 (2)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0179 (6)	0.0130 (6)	0.0113 (6)	0.0000 (5)	-0.0025 (5)	0.0028 (5)
C2	0.0164 (6)	0.0153 (6)	0.0136 (6)	0.0009 (5)	-0.0012 (5)	0.0028 (5)
C3	0.0202 (7)	0.0163 (6)	0.0147 (6)	0.0022 (5)	-0.0002 (5)	0.0004 (5)
C4	0.0180 (6)	0.0203 (7)	0.0171 (6)	-0.0029 (5)	-0.0012 (5)	0.0013 (5)
C5	0.0174 (7)	0.0160 (6)	0.0136 (6)	-0.0001 (5)	0.0007 (5)	0.0018 (5)
C6	0.0146 (6)	0.0135 (6)	0.0113 (6)	-0.0011 (5)	0.0014 (5)	0.0015 (5)
C7	0.0128 (6)	0.0164 (7)	0.0141 (6)	-0.0024 (5)	0.0012 (5)	-0.0004 (5)

C8	0.0169 (7)	0.0164 (6)	0.0146 (6)	-0.0010 (5)	-0.0031 (5)	0.0005 (5)
C9	0.0166 (7)	0.0182 (7)	0.0190 (7)	0.0013 (5)	-0.0023 (5)	-0.0009 (5)
C10	0.0189 (7)	0.0257 (7)	0.0178 (7)	-0.0006 (6)	0.0011 (5)	-0.0046 (6)
C11	0.0189 (7)	0.0247 (7)	0.0227 (7)	0.0018 (6)	0.0041 (6)	-0.0035 (6)
C12	0.0160 (6)	0.0228 (7)	0.0220 (7)	0.0016 (5)	0.0001 (5)	-0.0004 (6)
C13	0.0215 (7)	0.0240 (7)	0.0201 (7)	0.0010 (6)	0.0031 (6)	0.0033 (6)
C14	0.0208 (7)	0.0186 (7)	0.0278 (8)	-0.0017 (5)	0.0007 (6)	0.0002 (6)
C15	0.0200 (7)	0.0179 (7)	0.0192 (7)	-0.0008 (5)	-0.0015 (5)	-0.0034 (5)
C16	0.0203 (7)	0.0141 (6)	0.0192 (7)	0.0007 (5)	-0.0027 (5)	0.0002 (5)
C17	0.0174 (6)	0.0182 (7)	0.0177 (7)	0.0016 (5)	-0.0028 (5)	-0.0048 (5)
C18	0.0172 (6)	0.0197 (7)	0.0134 (6)	-0.0012 (5)	-0.0008 (5)	-0.0018 (5)
N1	0.0168 (5)	0.0131 (6)	0.0145 (5)	0.0014 (4)	-0.0039 (4)	-0.0029 (4)
N2	0.0234 (6)	0.0178 (6)	0.0161 (6)	-0.0027 (5)	-0.0025 (5)	-0.0003 (4)
N3	0.0159 (5)	0.0157 (5)	0.0127 (5)	-0.0020 (4)	-0.0007 (4)	-0.0027 (4)
O1	0.0165 (5)	0.0145 (5)	0.0166 (5)	0.0011 (3)	0.0001 (4)	-0.0001 (4)

Geometric parameters (Å, °)

C1—C2	1.3905 (19)	C11—C12	1.532 (2)
C1—C5	1.3950 (18)	C11—H11A	0.99
C1—C6	1.5026 (17)	C11—H11B	0.99
C2—C3	1.3875 (19)	C12—C13	1.5317 (19)
C2—H2	0.95	C12—H12A	0.99
C3—N2	1.3404 (18)	C12—H12B	0.99
C3—H3	0.95	C13—C14	1.529 (2)
C4—N2	1.3447 (19)	C13—H13A	0.99
C4—C5	1.3857 (19)	C13—H13B	0.99
C4—H4	0.95	C14—C15	1.533 (2)
C5—H5	0.95	C14—H14A	0.99
C6—O1	1.2304 (16)	C14—H14B	0.99
C6—N1	1.3567 (17)	C15—C16	1.5262 (18)
C7—N3	1.2853 (17)	C15—H15A	0.99
C7—C8	1.5091 (18)	C15—H15B	0.99
C7—C18	1.5144 (18)	C16—C17	1.5301 (19)
C8—C9	1.5414 (18)	C16—H16A	0.99
C8—H8A	0.99	C16—H16B	0.99
C8—H8B	0.99	C17—C18	1.5268 (19)
C9—C10	1.5286 (19)	C17—H17A	0.99
C9—H9A	0.99	C17—H17B	0.99
C9—H9B	0.99	C18—H18A	0.99
C10—C11	1.5354 (19)	C18—H18B	0.99
C10—H10A	0.99	N1—N3	1.3961 (15)
C10—H10B	0.99	N1—H1	0.88 (2)
C2—C1—C5	118.21 (12)	C13—C12—H12A	108.7
C2—C1—C6	123.96 (12)	C11—C12—H12A	108.7
C5—C1—C6	117.82 (12)	C13—C12—H12B	108.7
C3—C2—C1	118.59 (12)	C11—C12—H12B	108.7
C3—C2—H2	120.7	H12A—C12—H12B	107.6
C1—C2—H2	120.7	C14—C13—C12	114.23 (12)

N2—C3—C2	124.11 (12)	C14—C13—H13A	108.7
N2—C3—H3	117.9	C12—C13—H13A	108.7
C2—C3—H3	117.9	C14—C13—H13B	108.7
N2—C4—C5	123.89 (13)	C12—C13—H13B	108.7
N2—C4—H4	118.1	H13A—C13—H13B	107.6
C5—C4—H4	118.1	C13—C14—C15	114.11 (11)
C4—C5—C1	118.65 (12)	C13—C14—H14A	108.7
C4—C5—H5	120.7	C15—C14—H14A	108.7
C1—C5—H5	120.7	C13—C14—H14B	108.7
O1—C6—N1	124.31 (12)	C15—C14—H14B	108.7
O1—C6—C1	120.26 (11)	H14A—C14—H14B	107.6
N1—C6—C1	115.42 (11)	C16—C15—C14	114.11 (12)
N3—C7—C8	128.15 (12)	C16—C15—H15A	108.7
N3—C7—C18	116.66 (12)	C14—C15—H15A	108.7
C8—C7—C18	115.09 (11)	C16—C15—H15B	108.7
C7—C8—C9	111.48 (11)	C14—C15—H15B	108.7
C7—C8—H8A	109.3	H15A—C15—H15B	107.6
C9—C8—H8A	109.3	C15—C16—C17	114.22 (11)
C7—C8—H8B	109.3	C15—C16—H16A	108.7
C9—C8—H8B	109.3	C17—C16—H16A	108.7
H8A—C8—H8B	108	C15—C16—H16B	108.7
C10—C9—C8	113.16 (11)	C17—C16—H16B	108.7
C10—C9—H9A	108.9	H16A—C16—H16B	107.6
C8—C9—H9A	108.9	C18—C17—C16	113.74 (11)
C10—C9—H9B	108.9	C18—C17—H17A	108.8
C8—C9—H9B	108.9	C16—C17—H17A	108.8
H9A—C9—H9B	107.8	C18—C17—H17B	108.8
C9—C10—C11	113.64 (12)	C16—C17—H17B	108.8
C9—C10—H10A	108.8	H17A—C17—H17B	107.7
C11—C10—H10A	108.8	C7—C18—C17	117.13 (11)
C9—C10—H10B	108.8	C7—C18—H18A	108
C11—C10—H10B	108.8	C17—C18—H18A	108
H10A—C10—H10B	107.7	C7—C18—H18B	108
C12—C11—C10	114.72 (11)	C17—C18—H18B	108
C12—C11—H11A	108.6	H18A—C18—H18B	107.3
C10—C11—H11A	108.6	C6—N1—N3	116.91 (11)
C12—C11—H11B	108.6	C6—N1—H1	120.5 (12)
C10—C11—H11B	108.6	N3—N1—H1	120.4 (12)
H11A—C11—H11B	107.6	C3—N2—C4	116.42 (12)
C13—C12—C11	114.10 (12)	C7—N3—N1	118.17 (11)
C5—C1—C2—C3	2.74 (18)	C11—C12—C13—C14	-146.98 (12)
C6—C1—C2—C3	-178.29 (12)	C12—C13—C14—C15	68.58 (16)
C1—C2—C3—N2	0.7 (2)	C13—C14—C15—C16	67.05 (16)
N2—C4—C5—C1	1.9 (2)	C14—C15—C16—C17	-173.04 (11)
C2—C1—C5—C4	-3.94 (18)	C15—C16—C17—C18	70.36 (15)
C6—C1—C5—C4	177.03 (11)	N3—C7—C18—C17	33.46 (17)
C2—C1—C6—O1	-140.43 (13)	C8—C7—C18—C17	-149.87 (11)
C5—C1—C6—O1	38.54 (17)	C16—C17—C18—C7	64.33 (15)

C2—C1—C6—N1	41.05 (17)	O1—C6—N1—N3	-4.55 (19)
C5—C1—C6—N1	-139.98 (12)	C1—C6—N1—N3	173.91 (10)
N3—C7—C8—C9	-110.81 (15)	C2—C3—N2—C4	-2.77 (19)
C18—C7—C8—C9	72.97 (14)	C5—C4—N2—C3	1.44 (19)
C7—C8—C9—C10	64.83 (15)	C8—C7—N3—N1	-2.22 (19)
C8—C9—C10—C11	-173.28 (11)	C18—C7—N3—N1	173.95 (11)
C9—C10—C11—C12	68.04 (16)	C6—N1—N3—C7	-161.67 (12)
C10—C11—C12—C13	68.83 (16)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1 \cdots O1 ⁱ	0.88 (2)	2.15 (2)	3.0122 (15)	164.7 (16)

Symmetry code: (i) $-x+3/2, y+1/2, z$.